

JSS Academy of Higher Education and Research

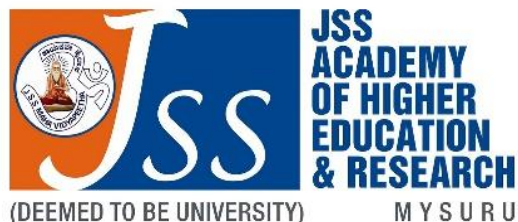
JSS College of Pharmacy

Sri Shivarathreshwara Nagara, Mysuru-570015

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Website: www.jssuni.edu.in

An ISO 9001:2015 Certified Institution



B. Pharm – I Semester Course Handout 2020-21



Accredited 'A+'
Grade by
NAAC



1st in Karnataka
& 3rd in INDIA to
be rated with 4
stars



Ranked 1st among
the YOUNG
UNIVERSITIES in
Karnataka



JSS College of
Pharmacy,
Mysuru – 10th
Rank in INDIA
2020



INTERNATIONAL
CERTIFICATION
Pharm D Program is
Certified by Accreditation
Council for Pharmacy
Education (ACPE), USA



ARIIA
ATAL RANKING OF INSTITUTIONS
ON INNOVATION ACHIEVEMENTS

Ranked 4th in India for 2019

Academic Calendar 2020-21 (B.Pharm - I Semester)

Teacher's Incharge

Class	Class Teacher	Batch No.	Batch Teacher
I B.Pharm I Semester	Dr. Sheshagiri Dixit	I	Dr. Sheshagiri Dixit
		II	Dr. Durai Anand Kumar
		III	Dr. B.M. Gurupadayya
		IV	Mrs. A.M. Mahalakshmi

ACTIVITIES AND COORDINATORS 2020-21

Curricular & Co curricular activities

Sl. No	Activities	Coordinator/s
1.	Induction, learning skills and personality development programs for fresher's	DHP/MPG
2.	Selection of class representative in first week of commencement of each course	
3.	Anti ragging cell	HP/ BM
4.	Grievance and redressal cell	PKK
5.	Industrial Visits, Training and placements	TS/ABP
6.	Guest lectures & Seminars/ conferences/ training / workshop <ul style="list-style-type: none"> • organized at college • delivered/attended by staff 	Respective department all HODs
7.	Internal Assessment Committee Chairperson Members	GVP RSS/SNM/DAT/BMV
8.	<ul style="list-style-type: none"> • Academic Council Board • Identification of Advanced/ Medium/ Slow learners 	Class Teachers Subject Teachers
9.	Ethics committee Meeting <ul style="list-style-type: none"> • Animal • Human 	KLK MR
10.	Time table	DHP TS/ URR/ VR/AMM/HYK
11.	Internal Quality Assurance Cell Chairperson Members	PKK/ AMM/AKT/HVG/SP

12.	Women's cell (Prevention of Sexual Harassment Cell)	SNM
13.	Scholarship Bureau	RSC
14.	Compilation of publications (Research papers/books/chapters)	BMG
15.	Research Coordination Committee -Compilation of Ph.D details and funded projects - Plagiarism - Review of publications	Chairperson – DVG Members – BRP/SB/JS
16.	Pharmacy Education Unit (CCLPE)	PKK/KU/RSS
17.	Annual result analysis List of merit students	UG – Subject Teacher, Class teacher & Program committee PG – Course Coordinator & Abhishek (Office)
18.	GPAT and other competitive exams (TOEFL, GRE etc.)	BM/ CSH/MPG
19.	Library orientation	Librarian
20.	Soft Skills Training	ABP

Extracurricular activities

Sl. No.	Activities	Coordinator/s
21.	<ul style="list-style-type: none"> • Selection of Class Representatives, Pharmaceutical society members • Annual planning and execution of Student centered and professional activities including inauguration of IPS 	MSS/ SRD
22.	JASPHARM	BS/ SM / CSH
23.	STUMAG	HYK
24.	Sports coordinators	MPV/HKS
25.	NSS coordinators	MPG / UM/ SND
26.	Cultural & Literary coordinators	KNS/CI

Other Institutional activities

Sl. No.	Activities	Coordinator/s
27.	Annual Day celebration / Graduation day	DAT/SM
28.	Course handouts/ Teachers diary/ Student handbook/Faculty handbook	HYK/PS
29.	National Pharmacy Week (NPW) & Pharmacists Day	VJ/ UM + IPA team
30.	Alumni association	HVG/ AKT/SM/BS

31.	Herbal and College Garden	JS/ NPK/VR
32.	ISO	DHP/SNM
33.	Press and publicity	KLK /BMV/OFFICE
34.	Foreign students cell	MPV
35.	Governing council meeting	JUS/ Office
36.	Monthly/Annual report of college activities to JSS AHER and other agencies	HoDs/JUS/ST/AKT/AM/KU/NPK Asha (office)
37.	College website	HKS/KU
38.	Research & Consultancy Co-ordinator • Collaboration with Industries/organizations • Interdepartment/Interdisciplinary research	DVG/ SB/ KM
39.	Coordinator - JSSUonline.com	ABP/TS
40.	JSSU Newsletter	KLK SRD/ KNS
41.	Annual group photo session	MSS/ SRD
42.	Lab coat and Blazers	JS / Ningaraju
43.	Notice Board (SNB, LNB and IIPC), Departmental staff list	Nagaraju
44.	Stock verification	Office staff /Librarian
45.	Student Liaison	Divya S
46.	Student ID Cards /Attendance entry	Shivanna / Manjunath
47.	Retreat for Pharmacy Students	AKT/ HKS/BRJ
48.	Feedback	VJ
49.	Institute Innovation Cell	HVG/PKK
50.	Practice School	MPG/VJ

Program Committee

Sl. No.	Program committees	Chairperson	Member Secretary
51.	D.Pharm	PKK	BMV
52.	B.Pharm	PKK	DAT
53.	Pharm.D	MR	RSS
54.	M.Pharm	PKK	SNM
55.	B.Pharm – Practice	MR	BRJ
56.	PG Diploma	PKK	JS

M.Pharm Program Coordinators

Sl. No.	M.Pharm Program	Coordinator
57.	Pharmaceutics	VJ

58.	Industrial Pharmacy	ABP
59.	Pharmaceutical Regulatory Affairs	MPV
60.	Pharmaceutical Quality Assurance	HVG
61.	Pharmaceutical Chemistry	BRP
62.	Pharmaceutical Analysis	BMG
63.	Pharmacology	KLK
64.	Pharmacognosy	NPK
65.	Pharmacy Practice	SP

PG Diploma Program Coordinators

Sl. No.	PG Diploma Program	Coordinator
66.	Pharmacovigilance	CSH
67.	Medicine & Poison Information	RSS
68.	Clinical Research	JUS
69.	Nanotechnology	VJ
70.	Pharmaceutical Quality Assurance	HVG
71.	Pharmaceutical Regulatory Affairs	MPV
72.	Medical Devices	BMV
73.	Intellectual Property Rights	BMV
74.	Computer Aided Drug Design	DAT
75.	Food and Drug Analysis	RSC
76.	Regulatory Toxicology	SB
77.	Phytopharmaceutical and Industrial Applications	JS

Certificate Course Coordinators

Sl. No.	Certificate Course	Coordinator
78.	Pharmaceutical Quality Assurance	HVG
79.	Herbal Drug Standardization	JS
80.	Medicine Information	RSS

TEACHING STAFF LIST

Sl. No	NAME	QUALIFICATION	DESIGNATION	Department
1	Dr. T.M. Pramod Kumar (TMP)	M.Pharm., Ph.D.	Professor & Principal	Pharmaceutics
2	Dr. P.K. Kulkarni (PKK)	M.Pharm., Ph.D.	Professor & Vice Principal	Pharmaceutics
3	Dr. D. Vishakante Gowda (DVG)	M.Pharm., Ph.D.	Professor & Head	Pharmaceutics
4	Dr. Balamuralidhara V. (BMV)	M.Pharm., Ph.D.	Asst. Professor	Pharmaceutics
5	Dr. Gangadharappa H.V.(HVG)	M.Pharm., Ph.D.	Asst. Professor	Pharmaceutics
6	Dr. M.P. Venkatesh (MPV)	M.Pharm., Ph.D.	Asst. Professor	Pharmaceutics
7	Dr. Vikas Jain (VJ)	M.Pharm., Ph.D.	Asst. Professor	Pharmaceutics
8	Dr. Amit B Patil (ABP)	M.Pharm., Ph.D.	Asst. Professor	Pharmaceutics
9	Dr. Gowrav M P (MPG)	M.Pharm., Ph.D.	Lecturer	Pharmaceutics
10	Mr. Hemanth Kumar S (HKS)	M.Pharm	Lecturer	Pharmaceutics
11	Mrs. Asha Spandana K M (ASP)	M.Pharm	Lecturer	Pharmaceutics
12	Mr B Mahendran (BM)	M.Pharm	Lecturer	Pharmaceutics
13	Dr Shailesh T (TS)	M.Pharm., Ph.D.	Lecturer	Pharmaceutics
14	Smt Preethi S (PS)	M.Pharm	Lecturer	Pharmaceutics
15	Dr. M. Ramesh (MR)	M.Pharm., Ph.D.	Professor & Head	Pharmacy Practice
16	Mr. D.H. P. Gowda (DHP)	M.Sc., PGDCA.	Asst. Professor	Pharmacy Practice
17	Mrs. Shilpa Palaksha (SP)	M.Pharm.	Asst. Professor	Pharmacy Practice
18	Mrs. Savitha R S (RSS)	M.Pharm.	Asst. Professor	Pharmacy Practice
19	Mr. Jaidev Kumar B R (BRJ)	M.Pharm.	Lecturer	Pharmacy Practice
20	Dr. M Umesh (UM)	Pharm D.	Lecturer	Pharmacy Practice
21	Dr. Juny Sebastian (JUS)	M.Pharm., Ph.D.	Lecturer	Pharmacy Practice
22	Dr Sri Harsha Chalasani (CSH)	M.Pharm., Ph.D.	Lecturer	Pharmacy Practice
23	Dr. Krishna Undela (KU)	M.Pharm., Ph.D.	Lecturer	Pharmacy Practice
24	Dr Srikanth M S (MSS)	M.Pharm., Ph.D.	Lecturer	Pharmacy Practice
25	Mr Balaji S (BS)	M.Pharm	Lecturer	Pharmacy Practice
26	Dr U R Rakshith (URR)	Pharm D	Lecturer	Pharmacy Practice
27	Dr. B.M. Gurupadayya (BMG)	M.Pharm., Ph.D.	Professor	Pharma. Chemistry
28	Dr. Gurubasavaraj V Pujar (GVP)	M.Pharm., Ph.D.	Professor & Head	Pharma. Chemistry
29	Dr. Prashantha Kumar B R (BRP)	M.Pharm., Ph.D.	Asst. Professor	Pharma. Chemistry
30	Dr. R. S. Chandan (RSC)	M.Pharm., Ph.D.	Asst. Professor	Pharma. Chemistry
31	Dr. Anand Kumar Tengli (AKT)	M.Pharm., Ph.D.	Asst. Professor	Pharma. Chemistry
32	Dr. Durai Ananda Kumar (DAT)	M.Pharm., Ph.D.	Asst. Professor	Pharma. Chemistry
33	Dr. Jaishree V (JV)	M.Pharm., Ph.D.	Asst. Professor	Pharma. Chemistry
34	Dr. H. Yogish Kumar (HYK)	M.Pharm., Ph.D.	Lecturer	Pharma. Chemistry

35	Dr. Sheshagiri Dixit (SRD)	M.Pharm., Ph.D.	Lecturer	Pharma. Chemistry
36	Mr. Chetan.IA	M.Pharm	Lecturer	Pharma. Chemistry
37	Dr. K Mruthunjaya (KM)	M.Pharm., Ph.D.	Professor & Head	Pharmacognosy
38	Dr. J. Suresh (JS)	M.Pharm., Ph.D.	Professor	Pharmacognosy
39	Dr. N Paramakrishnan (NPK)	M.Pharm., Ph.D.	Lecturer	Pharmacognosy
40	Mr. Vageesh Revadigar (VR)	M.Pharm	Lecturer	Pharmacognosy
41	Ms. HariPriya G	M Pharm	Lecturer	Pharmacognosy
42	Dr. S. N. Manjula (SNM)	M.Pharm., Ph.D.	Professor & Head	Pharmacology
43	Dr. Saravana Babu C (SB)	M.Pharm., Ph.D.	Asso.Professor	Pharmacology
44	Dr. K L Krishna (KLK)	M.Pharm., Ph.D.	Asst. Professor	Pharmacology
45	Mrs. A M Mahalakshmi (AMM)	M.Pharm.	Lecturer	Pharmacology
46	Mrs. Seema Mehdi (SM)	M.Pharm	Lecturer	Pharmacology
47	Dr. Nagashree K S (KNS)	M.Pharm., Ph.D	Lecturer	Pharmacology

B.PHARM**Program Educational Objectives (PEOs):**

PEO 1: To acquire the theoretical knowledge of pharmaceutical sciences

PEO 2: To acquire practical skills in

- isolation of medicinal compounds from natural sources
- synthesis and analysis of medicinal compounds
- screening medicinal compounds for pharmacological activities
- formulation of pharmaceutical dosage forms and their evaluation

PEO 3: To develop competent Pharmacists with ethical attitude, research intuition, leadership qualities, to participate in public health programs and engage in life-long learning

Program Outcomes (POs):

1. Ability to acquire knowledge of pharmaceutical sciences
2. Ability to design and conduct experiments, to analyze and interpret data
3. Ability to demonstrate effective planning, develop and implement plans within time frame.
4. Ability to function effectively individually and on teams, including diverse and multidisciplinary, to accomplish a task.
5. Ability to understand and appreciate the role of pharmacist in healthcare services.
6. Understanding of professional, ethical, legal, security and social issues and responsibilities.
7. Ability to understand contemporary issues relating to pharmacy profession and challenges ahead.
8. Awareness of ethical and professional responsibilities.
9. Possess the necessary interpersonal and communication skills to be a productive member of the team in work environment.
10. Ability to use current techniques, skills, and modern tools.
11. A strong background and motivation to pursue life-long learning

COURSE HANDOUT 2020-21**Class: B. Pharm - I Semester****1. Course Details**

Course code	Name of the course	No. of hours	Tutorial	Credit points
BP101T	Human Anatomy and Physiology I– Theory	3	1	4
BP102T	Pharmaceutical Analysis – Theory	3	1	4
BP103T	Pharmaceutics I – Theory	3	1	4
BP104T	Pharmaceutical Inorganic Chemistry – Theory	3	1	4
BP105T	Communication skills – Theory *	2	-	2
BP106RBT BP106RMT	Remedial Biology/ Mathematics – Theory*	2	-	2
BP107P	Human Anatomy and Physiology I– Practical	4	-	2
BP108P	Pharmaceutical Analysis I – Practical	4	-	2
BP109P	Pharmaceutics I – Practical	4	-	2
BP110P	Pharmaceutical Inorganic Chemistry – Practical	4	-	2
BP111P	Communication skills – Practical*	2	-	1
BP112RBP	Remedial Biology – Practical*	2	-	1
Total		32/34^{\$}/36[#]	4	27/29^{\$}/30[#]

^{\$}Applicable ONLY for the students studied Physics / Chemistry / Botany / Zoology at HSC and appearing for Remedial Mathematics course

[#] Applicable ONLY for the students appearing for Remedial Biology / Mathematics

* Non University Examination (NUE)

2. Evaluation:**a. Internal assessment: Continuous mode**

The marks allocated for Continuous mode of Internal Assessment, as per the scheme given below.

Table 1: Scheme for awarding internal assessment: Continuous mode

THEORY		
Criteria	Maximum Marks	
Attendance	4	2
Academic activities (Average of any 3 activities e.g. quiz, assignment, open book test, field work, group discussion and seminar)	3	1.5
Student – Teacher interaction	3	1.5
Total	10	5
PRACTICALS		
Attendance	2	
Based on Practical Records, Regular viva voce, etc.	3	
Total	5	

Table 2: Guidelines for the allotment of marks for attendance

Percentage of Attendance	Theory	Practical
95 – 100	4	2
90 – 94	3	1.5
85 – 89	2	1
80 – 84	1	0.5
Less than 80	0	0

b. Sessional Exams

Two Sessional exams shall be conducted for each theory / practical course as per the schedule fixed by the college(s). The scheme of question paper for theory and practical Sessional examinations is given below. The average marks of two Sessional exams shall be computed for internal assessment as per the requirements.

Sessional exam shall be conducted for 30 marks for theory and shall be computed for 15 marks. Similarly Sessional exam for practical shall be conducted for 40 marks and shall be computed for 10 marks.

Question paper pattern for theory Sessional examinations**For subjects having University examination**

I. Multiple Choice Questions (MCQs) (Answer all the questions)	=	10 x 1 = 10
I. Long Answers (Answer 1 out of 2)	=	1 x 10 = 10
II. Short Answers (Answer 2 out of 3)	=	2 x 5 = 10

Total	=	30 marks

For subjects having Non University Examination

I. Long Answers (Answer 1 out of 2)	=	1 x 10 = 10
II. Short Answers (Answer 4 out of 6)	=	4 x 5 = 20

Total	=	30 marks

Question paper pattern for practical sessional examinations

I. Synopsis	=	10
II. Experiments	=	25
III. Viva voce	=	05

Total	=	40 marks

3. End semester examinations

The End Semester Examinations for each theory and practical course through semesters I to VIII shall be conducted by the university except for the subjects notified as non-university examinations

Table 3: Scheme for internal assessments and university examination - Semester-I

Course code	Name of the course	Internal Assessment				University Exam		Total Marks
		Continuo us Mode	Sessional Exams		Total	Marks	Duration	
			Marks	Duration				
BP101T	Human Anatomy and Physiology I– Theory	10	15	1 Hr	25	75	3 Hrs	100
BP102T	Pharmaceutical Analysis I – Theory	10	15	1 Hr	25	75	3 Hrs	100
BP103T	Pharmaceutics I – Theory	10	15	1 Hr	25	75	3 Hrs	100
BP104T	Pharmaceutical Inorganic Chemistry – Theory	10	15	1 Hr	25	75	3 Hrs	100
BP105T	Communication skills – Theory *	5	10	1 Hr	15	35	1.5 Hrs	50
BP106RBT BP106RMT	Remedial Biology/Mathematics – Theory*	5	10	1 Hr	15	35	1.5 Hrs	50
BP107P	Human Anatomy and Physiology – Practicals	5	10	4 Hrs	15	35	4 Hrs	50
BP108P	Pharmaceutical Analysis I – Practicals	5	10	4 Hrs	15	35	4 Hrs	50
BP109P	Pharmaceutics I – Practicals	5	10	4 Hrs	15	35	4 Hrs	50
BP110P	Pharmaceutical Inorganic Chemistry – Practicals	5	10	4 Hrs	15	35	4 Hrs	50
BP111P	Communication skills – Practicals*	5	5	2 Hrs	10	15	2 Hrs	25
BP112RBP	Remedial Biology – Practicals*	5	5	2 Hrs	10	15	2 Hrs	25
Total		70/75[§]/80[#]	115/125[§]/130[#]	23/24[§]/26[#] Hrs	185/200[§]/210[#]	490/525[§]/540[#]	31.5/33[§]/35[#] Hrs	675/725[§]/750[#]

Applicable ONLY for the students studied Mathematics / Physics / Chemistry at HSC and appearing for Remedial Biology (RB) course

§ Applicable ONLY for the students studied Physics / Chemistry / Botany / Zoology at HSC and appearing for Remedial Mathematics (RM) course

* Examinations shall be conducted by the subject experts at college level

4. Promotion and award of grades

A student shall be declared PASS and eligible for getting grade in a course of B.Pharm. programme if he/she secures at least 50% marks in that particular course including internal assessment. For example, to be declared as PASS and to get grade, the student has to secure a minimum of 50 marks for the total of 100 including continuous mode of assessment and end

semester theory examination and has to secure a minimum of 25 marks for the total 50 including internal assessment and end semester practical examination.

5. Carry forward of marks

In case a student fails to secure the minimum 50% in any Theory or Practical course as specified (in promotion and award of grades), then he/she shall reappear for the university examination of that course. However his/her marks of the Internal Assessment shall be carried over and he/she shall be entitled for grade obtained by him/her on passing.

6. Improvement of internal assessment

A student shall have the opportunity to improve his/her performance only once in the sessional exam component of the Internal assessment. The re-conduct of the sessional exam should be completed before the commencement of next semester theory examinations.

7. Re-examination of end semester examinations

Reexamination of end semester examination shall be conducted as per the schedule given in table 3. The exact dates of examinations will be notified from time to time.

Table 4: Tentative schedule of university examinations and supplementary examinations

Semester	Regular examinations	Supplementary examinations
I, III, V and VII	November / December	May / June
II, IV, VI and VIII	May / June	November / December

Question pattern for university theory examinations for 75 marks paper

I. Multiple Choice Questions (MCQs) (Answer all the questions)	=	20 x 1 = 20
I. Long Answers (2 out of 3)	=	2 x 10 = 20
II. Short Answers (7 out of 9)	=	7 x 5 = 35

Total	=	75 marks

Question pattern for university theory examinations for 50 marks paper

I. Long Answers (2 out of 3)	=	2 x 10 = 20
II. Short Answers (6 out of 8)	=	6 x 5 = 30

Total	=	50 marks

8. Grading of performances

Letter grades and grade points allocations

Based on the performances, each student shall be awarded a final letter grade at the end of the semester for each course.

Table 5: Letter grades and grade points equivalent to percentage of marks and performances

Percentage of Marks Obtained	Letter Grade	Grade Point	Performance
90.00 – 100	A+	10	Outstanding
80.00 – 89.99	A	9	Excellent
70.00 – 79.99	B	8	Good
60.00 – 69.99	C	7	Fair
50.00 – 59.99	D	6	Average
Less than 50	F	0	Fail
Absent	AB	0	Fail

A learner who remains absent in any form of evaluation/examination, letter grade allocated to him/her should be assigned a letter grade of AB and a corresponding grade point of zero. He/she should reappear for the said evaluation/examination in due course.

9. Declaration of class

The class shall be awarded on the basis of CGPA as follows:

First Class with Distinction	= CGPA of 7.50 and above
First Class	= CGPA of 6.00 to 7.49
Second Class	= CGPA of 5.00 to 5.99

10. Attendance: The marks is allotted based on the attendance percentage (Table 2)

11. Chamber consultation hours: Any time during college hours.

12. Tutorial Class: Objective of the tutorial is to enhance the learning ability and help students in better understanding of the subject. This provides a best opportunity for the students to clarify their subject doubts. This involves discussions, presentations on specified topics, assignments and evaluation.

BP101T- HUMAN ANATOMY AND PHYSIOLOGY-I (THEORY)

Teacher/s: Mrs. A.M. Mahalakshmi (AMM)

45 Hours (3 Hrs/ week)

Scope: This subject is designed to impart fundamental knowledge on the structure and functions of the various systems of the human body. It also helps in understanding both homeostatic mechanisms. The subject provides the basic knowledge required to understand the various disciplines of pharmacy.

Objectives: Upon completion of this course the student should be able to

- Explain the gross morphology, structure and functions of various organs of the human body.
- Describe the various homeostatic mechanisms and their imbalances.
- Identify the various tissues and organs of different systems of human body.
- Perform the various experiments related to special senses and nervous system.
- Appreciate coordinated working pattern of different organs of each system

Lecture wise programme:

Topic	Hrs
<p>1. Introduction to human body Definition and scope of anatomy and physiology, levels of structural organization and body systems, basic life processes, homeostasis, basic anatomical terminology.</p>	03
<p>2. Cellular level of organization Structure and functions of cell, transport across cell membrane, cell division, cell junctions. General principles of cell communication, intracellular signaling pathway activation by extracellular signal molecule, Forms of intracellular signaling: a) Contact-dependent b) Paracrine c) Synaptic d) Endocrine</p>	04
<p>3. Tissue level of organization Classification of tissues, structure, location and functions of epithelial, muscular and nervous and connective tissue</p>	03
<p>4. Integumentary system Structure and functions of skin</p>	02
<p>5. Skeletal system Divisions of skeletal system, types of bone, salient features and functions of bones of axial and appendicular skeletal system Organization of skeletal muscle, physiology of muscle contraction, neuromuscular junction</p>	06

6.	Joints Structural and functional classification, types of movements	02
7.	Body fluids and blood Body fluids, composition and functions of blood, hemopoiesis, formation of hemoglobin, anemia, mechanisms of coagulation, blood grouping, Rh factors, transfusion, its significance and disorders of blood, Reticulo endothelial system	05
8.	Lymphatic system Lymphatic organs and tissues, lymphatic vessels, lymph circulation and functions of lymphatic system	05
9.	Peripheral nervous system Classification of peripheral nervous system: Structure and functions of sympathetic and parasympathetic nervous system. Origin and functions of spinal and cranial nerves.	04
10.	Special senses Structure and function of eye, ear, nose and tongue and their disorders	04
11.	Cardiovascular system Heart – anatomy of heart, blood circulation, blood vessels, structure and functions of artery, vein and capillaries, elements of conduction system of heart and heart beat, its regulation by autonomic nervous system, cardiac output, cardiac cycle. Regulation of blood pressure, pulse, electrocardiogram and disorders of heart.	07

Theory Sessional examination syllabus

Sessional No.	Syllabus
	Chapters no.
I	1 to 7
II	8 to 11

BP107P-HUMAN ANATOMY AND PHYSIOLOGY-I (PRACTICALS)**Teacher: Mrs. A.M. Mahalakshmi (AMM)****60 Hours (4 Hrs/week)****Title of Experiments**

1. Study of compound microscope.
2. Microscopic study of epithelial and connective tissue
3. Microscopic study of muscular and nervous tissue
4. Identification of axial bones
5. Identification of appendicular bones
6. Introduction to hemocytometry.
7. Enumeration of white blood cell (WBC) count
8. Enumeration of total red blood corpuscles (RBC) count
9. Determination of bleeding time
10. Determination of clotting time
11. Estimation of hemoglobin content
12. Determination of blood group.
13. Determination of erythrocyte sedimentation rate (ESR).
14. Determination of heart rate and pulse rate.
15. Recording of blood pressure.

Recommended Books (Latest Editions)

1. Essentials of Medical Physiology by K. Sembulingam and P. Sembulingam. Jaypee
2. brothers medical publishers, New Delhi.
3. Anatomy and Physiology in Health and Illness by Kathleen J.W. Wilson, Churchill
4. Livingstone, New York
5. Physiological basis of Medical Practice-Best and Tailor. Williams & Wilkins
6. Co,Riverview,MI USA
7. Text book of Medical Physiology- Arthur C,Guyton andJohn.E. Hall. Miamisburg, OH,
8. U.S.A.
9. Principles of Anatomy and Physiology by Tortora Grabowski. Palmetto, GA, U.S.A.
10. Textbook of Human Histology by Inderbir Singh, Jaypee brother's medical publishers,
11. New Delhi.
12. Textbook of Practical Physiology by C.L. Ghai, Jaypee brother's medical publishers,
13. New Delhi.

14. Practical workbook of Human Physiology by K. Srinageswari and Rajeev Sharma,
15. Jaypee brother's medical publishers, New Delhi.

Reference Books (Latest Editions)

1. Physiological basis of Medical Practice-Best and Tailor. Williams & Wilkins Co, Riverview, MI USA
2. Text book of Medical Physiology- Arthur C, Guyton and John. E. Hall. Miamisburg, OH, U.S.A.
3. Human Physiology (vol 1 and 2) by Dr. C.C. Chatterje ,Academic Publishers Kolkata

BP102T- PHARMACEUTICAL ANALYSIS (THEORY)

Teacher/s: Dr. Sheshagiri Dixit (SRD)

45 Hours (3 Hrs/ week)

Scope: This course deals with the fundamentals of analytical chemistry and principles of electrochemical analysis of drugs

Objectives: Upon completion of the course student shall be able to

- understand the principles of volumetric and electro chemical analysis
- carryout various volumetric and electrochemical titrations
- develop analytical skills

Course Materials

1. A.H. Beckett & J.B. Stenlake's, Practical Pharmaceutical Chemistry Vol I & II, Stahlone Press of University of London, 4th edition.
2. A.I. Vogel, Text Book of Quantitative Inorganic analysis
3. P. Gundu Rao, Inorganic Pharmaceutical Chemistry, 3rd Edition
4. Bentley and Driver's Textbook of Pharmaceutical Chemistry
5. John H. Kennedy, Analytical chemistry principles
6. Indian Pharmacopoeia.

Lecture wise Programme

TOPICS	
1. Pharmaceutical analysis: Definition and scope i. Different techniques of analysis ii. Methods of expressing concentration iii. Primary and secondary standards iv. Preparation and standardization of various molar and normal solutions- Oxalic acid, sodium hydroxide, hydrochloric acid, sodium thiosulphate, sulphuric acid, potassium permanganate and ceric ammonium sulphate	02 02 02 02
2. Errors: Sources of errors, types of errors, methods of minimizing errors, accuracy, precision and significant figures	02
3. Acid base titration: Theories of acid base indicators, classification of acid base titrations and theory involved in titrations of strong, weak, and very	06

weak acids and bases, neutralization curves	
4. Non aqueous titration: Solvents, acidimetry and alkalimetry titration and estimation of Sodium benzoate and Ephedrine HCl	04
5. Precipitation titrations: Mohrs method, Volhard's, Modified Volhard's, Fajans method, estimation of sodium chloride.	04
6. Complexometric titration: Classification, metal ion indicators, masking and demasking reagents, estimation of Magnesium sulphate, and calcium gluconate	03 03
7. Gravimetry: Principle and steps involved in gravimetric analysis. Purity of the precipitate: co-precipitation and post precipitation, Estimation of barium sulphate. Basic Principles, methods and application of diazotisation titration.	
Redox titrations	
8. Concepts of oxidation and reduction	03
9. Types of redox titrations (Principles and applications) Cerimetry, Iodimetry, Iodometry, Bromatometry, Dichrometry, Titration with potassium iodate	05
Electrochemical methods of analysis	
10. Conductometry - Introduction, Conductivity cell, Conductometric titrations, applications.	02
11. Potentiometry - Electrochemical cell, construction and working of reference (Standard hydrogen, silver chloride electrode and calomel electrode) and indicator electrodes (metal electrodes and glass electrode), methods to determine end point of potentiometric titration and applications.	03 02
12. Polarography - Principle, Ilkovic equation, construction and working of dropping mercury electrode and rotating platinum electrode, applications	

Theory Internal assessment syllabus

Internal assessment No.	Syllabus
	Chapters no.
I	1 – 4, 6
II	5, 7 - 12

BP108P - PHARMACEUTICAL ANALYSIS (PRACTICALS)**Teacher/s: Dr. Sheshagiri Dixit (SRD)****60 Hours (4 Hrs/ week)****General Requirements:** Graph paper, pencil, Scale, Scissors, Butter Paper,

Observation Book-75 pages (plain)

Gum Tube or stick, Matchbox, Laboratory Napkin

List of Experiments:**I Limit Test of the following**

- (1) Chloride
- (2) Sulphate
- (3) Iron
- (4) Arsenic

II Preparation and standardization of

- (1) Sodium hydroxide
- (2) Sulphuric acid
- (3) Sodium thiosulfate
- (4) Potassium permanganate
- (5) Ceric ammonium sulphate

III Assay of the following compounds along with Standardization of Titrant

- (1) Ammonium chloride by acid base titration
- (2) Ferrous sulphate by Cerimetry
- (3) Copper sulphate by Iodometry
- (4) Calcium gluconate by complexometry
- (5) Hydrogen peroxide by Permanganometry
- (6) Sodium benzoate by non-aqueous titration
- (7) Sodium Chloride by precipitation titration

IV Determination of Normality by electro-analytical methods

- (1) Conductometric titration of strong acid against strong base
- (2) Conductometric titration of strong acid and weak acid against strong base
- (3) Potentiometric titration of strong acid against strong base

Recommended Books: (Latest Editions)

1. A.H. Beckett & J.B. Stenlake's, Practical Pharmaceutical Chemistry Vol I & II, Stahlone Press of University of London
2. A.I. Vogel, Text Book of Quantitative Inorganic analysis
3. P. Gundu Rao, Inorganic Pharmaceutical Chemistry
4. Bentley and Driver's Textbook of Pharmaceutical Chemistry
5. John H. Kennedy, Analytical chemistry principles
6. Indian Pharmacopoeia

BP103T – PHARMACEUTICS -I (THEORY)**Teacher/s: Dr. Shailesh T (TS) & Mr. B. Mahendran (MB)****45 Hours (3 Hrs/ week)**

Scope: This course is designed to impart a fundamental knowledge on the preparatory pharmacy with art and science of preparing the different conventional dosage forms.

Objectives: Upon completion of this course the student should be able to:

- Know the history of profession of pharmacy
- Understand the basics of different dosage forms, pharmaceutical incompatibilities and pharmaceutical calculations
- Understand the professional way of handling the prescription
- Preparation of various conventional dosage forms

Course Content:

- 1. Historical background and development of profession of pharmacy: 04 Hours**
History of profession of Pharmacy in India in relation to pharmacy education, industry and organization, Pharmacy as a career, Pharmacopoeias: Introduction to IP, BP, USP and Extra Pharmacopoeia.
- 2. Dosage forms: Introduction to dosage forms, classification and definitions 02 Hours**
- 3. Prescription: Definition, Parts of prescription, handling of Prescription and Errors in prescription. 04 Hours**
Posology: Definition, Factors affecting posology. Pediatric dose calculations based on age, body weight and body surface area.
- 4. Pharmaceutical calculations: Weights and measures – Imperial & Metric system, Calculations involving percentage solutions, alligation, proof spirit and isotonic solutions based on freezing point and molecular weight. 04 Hours**
- 5. Powders: Definition, classification, advantages and disadvantages, Simple & compound powders – official preparations, dusting powders, effervescent, efflorescent and hygroscopic powders, eutectic mixtures. Geometric dilutions 04 Hours**
- 6. Liquid dosage forms: Advantages & Disadvantages of liquid dosage forms. Excipients used in formulation of liquid dosage forms. Solubility enhancement techniques 02 Hours**

a. Monophasic liquids: Definitions and preparations of Gargles, Mouthwashes, Throat Paint, Eardrops, Nasal drops, Enemas, Syrups, Elixirs, Liniments and Lotions. **02 Hours**

b. Biphasic liquids:

Suspensions: Definition, advantages and disadvantages, classifications.

Preparation of suspensions; Flocculated and Deflocculated suspension, stability problems and methods to overcome. **04 Hours**

Emulsions: Definition, classification, emulsifying agent, test for the identification of type of Emulsion, Methods of preparation & stability problems and methods to overcome.

04 Hours

7. Semisolid dosage forms: Definitions, classification, mechanisms and factors influencing dermal penetration of drugs. Preparation of ointments, pastes, creams and gels. Excipients used in semi solid dosage forms. Evaluation of semi solid dosages forms **07 Hours**

Recommended Books: (Latest Editions)

1. H.C. Ansel et al., Pharmaceutical Dosage Form and Drug Delivery System, Lippincott Williams and Walkins, New Delhi.
2. Carter S.J., Cooper and Gunn's-Dispensing for Pharmaceutical Students, CBS publishers, New Delhi.
3. M.E. Aulton, Pharmaceutics, The Science & Dosage Form Design, Churchill Livingstone, Edinburgh.
4. Indian pharmacopoeia.
5. British pharmacopoeia.
6. Lachmann. Theory and Practice of Industrial Pharmacy, Lea & Febiger Publisher, The University of Michigan.
7. Alfonso R. Gennaro Remington. The Science and Practice of Pharmacy, Lippincott Williams, New Delhi.
8. Carter S.J., Cooper and Gunn's. Tutorial Pharmacy, CBS Publications, New Delhi.
9. E.A. Rawlins, Bentley's Text Book of Pharmaceutics, English Language Book Society, Elsevier Health Sciences, USA.
10. Isaac Ghebre Sellassie: Pharmaceutical Pelletization Technology, Marcel Dekker, INC, New York.
11. Dilip M. Parikh: Handbook of Pharmaceutical Granulation Technology, Marcel Dekker, INC, New York.
12. Françoise Nieloud and Gilberte Marti-Mestres: Pharmaceutical Emulsions and Suspensions, Marcel Dekker, INC, New York.

Theory Internal assessment syllabus

Internal assessment No.	Syllabus
	Chapters no.
I	1 – 6a
II	6b - 9

BP109P - PHARMACEUTICS I (PRACTICALS)**Teacher/s: Dr. Shailesh T (ST) & Mr. Mahendran B (MB)****60 Hours (4 Hrs/week)****1. Syrups**

- a) Syrup IP'66
- b) Compound syrup of Ferrous Phosphate BPC'68

2. Elixirs

- a) Piperazine Citrate elixir
- b) Paracetamol pediatric elixir

3. Linctus

- a) Terpin Hydrate Linctus IP'66
- b) Iodine Throat Paint (Mandles Paint)

4. Solutions

- a) Strong solution of ammonium acetate
- b) Cresol with soap solution
- c) Lugol's solution

5. Suspensions

- a) Calamine lotion
- b) Magnesium Hydroxide mixture
- c) Aluminium Hydroxide gel

6. Emulsions

- a) Turpentine Liniment
- b) Liquid paraffin emulsion

7. Powders and Granules

- a) ORS powder (WHO)
- b) Effervescent granules
- c) Dusting powder
- d) Divded powders

8. Suppositories

- a) Glycero gelatin suppository
- b) Coca butter suppository
- c) Zinc Oxide suppository

8. Semisolids

- a) Sulphur ointment
- b) Non staining-iodine ointment with methyl salicylate
- c) Carbopal gel

9. Gargles and Mouthwashes

- a) Iodine gargle
- b) Chlorhexidine mouthwash

BP104T - PHARMACEUTICAL INORGANIC CHEMISTRY (THEORY)**Teacher/s: Dr. T. Durai Anand Kumar (DAK)****45 Hrs (3 Hrs/week)****Scope:** This subject deals with the monographs of inorganic drugs and pharmaceuticals.**Objectives:** Upon completion of course student shall be able to

- know the sources of impurities and methods to determine the impurities in inorganic drugs and pharmaceuticals
- understand the medicinal and pharmaceutical importance of inorganic compounds

Lecture wise Programme:

	Topic	Hrs
1.	Impurities in pharmaceutical substances: History of Pharmacopoeia, Sources and types of impurities, principle involved in the limit test for Chloride, Sulphate, Iron, Arsenic, Lead and Heavy metals, modified limit test for Chloride and Sulphate	10
	General methods of preparation, assay for the compounds superscripted with asterisk (*), properties and medicinal uses of inorganic compounds belonging to the following classes	
2.	Acids, Bases and Buffers: Buffer equations and buffer capacity in general, buffers in pharmaceutical systems, preparation, stability, buffered isotonic solutions, measurements of tonicity, calculations and methods of adjusting isotonicity. Major extra and intracellular electrolytes: Functions of major physiological ions, Electrolytes used in the replacement therapy: Sodium chloride*, Potassium chloride, Calcium gluconate* and Oral Rehydration Salt (ORS), Physiological acid base balance. Dental products: Dentifrices, role of fluoride in the treatment of dental caries, Desensitizing agents, Calcium carbonate, Sodium fluoride, and Zinc eugenol cement.	03
		04
		03
3.	Gastrointestinal agents	
	a. Acidifies: Ammonium chloride* and Dil. HCl	01
	b. Antacid: Ideal properties of antacids, combinations of antacids, Sodium Bicarbonate*, Aluminum hydroxide gel, Magnesium hydroxide mixture	03 02
	c. Cathartics: Magnesium sulphate, Sodium orthophosphate, Kaolin and Bentonite	
	d. Antimicrobials: Mechanism, classification. Potassium permanganate, Boric acid, Hydrogen peroxide*, Chlorinated lime*, Iodine and its preparations	04

4.	Miscellaneous compounds	
	a. Expectorants: Potassium iodide, Ammonium chloride*.	02
	b. Emetics: Copper sulphate*, Sodium potassium tartarate	02
	c. Haematinics: Ferrous sulphate*, Ferrous gluconate	01
	d. Poison and Antidote: Sodium thiosulphate*, Activated charcoal, Sodium nitrite	02
	e. Astringents: Zinc Sulphate, Potash Alum	01
5.	Radiopharmaceuticals	07
	Radio activity, Measurement of radioactivity, Properties of α , β , γ radiations, Half life, radio isotopes and study of radio isotopes - Sodium iodide I ¹²¹ , Storage conditions, precautions & pharmaceutical application of radioactive substances.	

Recommended Books (Latest Editions)

1. A.H. Beckett & J.B. Stenlake's, Practical Pharmaceutical Chemistry Vol I & II, Stahlone Press of University of London, 4th edition.
2. A.I. Vogel, Text Book of Quantitative Inorganic analysis
3. P. Gundu Rao, Inorganic Pharmaceutical Chemistry, 3rd Edition
4. M.L Schroff, Inorganic Pharmaceutical Chemistry
5. Bentley and Driver's Textbook of Pharmaceutical Chemistry
6. Anand & Chatwal, Inorganic Pharmaceutical Chemistry
7. Indian Pharmacopoeia

Theory Sessional examination syllabus

Sessional No.	Syllabus Chapter No.
I	1 to 3c
II	3d to 5

BP110P - PHARMACEUTICAL INORGANIC CHEMISTRY (PRACTICAL)**Teacher/s: Dr. T. Durai Anand Kumar (DAK)****60 Hours (4 Hrs / Week)****I Limit tests for following ions**

1. Limit test for Chlorides and Sulphates
2. Modified limit test for Chlorides and Sulphates
3. Limit test for Iron
4. Limit test for Heavy metals
5. Limit test for Lead
6. Limit test for Arsenic

II Identification test

1. Magnesium hydroxide
2. Ferrous sulphate
3. Sodium bicarbonate
4. Calcium gluconate
5. Copper sulphate

III Test for purity

1. Swelling power of Bentonite
2. Neutralizing capacity of aluminum hydroxide gel
3. Determination of potassium iodate and iodine in potassium Iodide

IV Preparation of inorganic pharmaceuticals

1. Boric acid
2. Potash alum
3. Ferrous sulphate

BP105T - COMMUNICATIONSKILLS (THEORY)

Teacher/s: Arivu Team

30 Hrs (2 Hrs/week)

Scope: This course will prepare the young pharmacy student to interact effectively with doctors, nurses, dentists, physiotherapists and other health workers. At the end of this course the student will get the soft skills set to work cohesively with the team as a team player and will add value to the pharmaceutical business.

Objectives:

Upon completion of the course the student shall be able to

1. understand the behavioral needs for a Pharmacist to function effectively in the areas of pharmaceutical operation
2. communicate effectively (Verbal and Non Verbal)
3. effectively manage the team as a team player
4. develop interview skills
5. develop Leadership qualities and essentials

Lecture wise Programme:

	Topic	Hrs
1.	Communication Skills Introduction, Definition, The Importance of Communication The Communication Process – Source, Message, Encoding, Channel, Decoding, Receiver, Feedback, Context	03
2.	Barriers to communication Physiological Barriers, Physical Barriers, Cultural Barriers, Language Barriers, Gender Barriers, Interpersonal Barriers, Psychological Barriers, Emotional barriers	02
3.	Perspectives in Communication Introduction, Visual Perception, Language, Other factors affecting our perspective - Past Experiences, Prejudices, Feelings, Environment	03
4.	Elements of Communication Introduction, Face to Face Communication - Tone of Voice, Body Language (Non-verbal communication), Verbal Communication, Physical Communication	05
5.	Communication Styles Introduction, The Communication Styles Matrix with example for each -Direct Communication Style, Spirited Communication Style, Systematic Communication Style, Considerate Communication Style	02
6.	Basic Listening Skills Introduction, Self-Awareness, Active Listening, Becoming an Active Listener, Listening in Difficult Situations	03
7.	Effective Written Communication	04

	Introduction, When and When Not to Use Written Communication - Complexity of the Topic, Amount of Discussion' Required, Shades of Meaning, Formal Communication Writing Effectively: Subject Lines, Put the Main Point First, Know Your audience, Organization of the Message	
8.	Interview Skills Purpose of an interview, Do's and Dont's of an interview	02
9.	Giving Presentations Dealing with Fears, Planning your Presentation, Structuring Your Presentation, Delivering Your Presentation, Techniques of Delivery	03
10.	Group Discussion Introduction, Communication skills in group discussion, Do's and Dont's of group discussion	03

BP111P - COMMUNICATION SKILLS (PRACTICALS)

Teacher/s: Arivu Team

30 Hrs (2 Hrs/week)

The following learning modules are to be conducted using wordsworth® english language lab software

Basic communication covering the following topics

1. Meeting People
2. Asking Questions
3. Making Friends
4. What did you do?
5. Do's and Dont's

Pronunciations covering the following topics

6. Pronunciation (Consonant Sounds)
7. Pronunciation and Nouns
8. Pronunciation (Vowel Sounds)

Advanced Learning covering the following topics

9. Listening Comprehension / Direct and Indirect Speech
10. Figures of Speech
11. Effective Communication
12. Writing Skills
13. Effective Writing
14. Interview Handling Skills

15. E-Mail etiquette
16. Presentation Skills

Recommended Books

1. Basic communication skills for Technology, Andreja. J. Ruther Ford, 2nd Edition, Pearson Education, 2011
2. Communication skills, Sanjay Kumar, Pushpalata, 1st Edition, Oxford Press, 2011
3. Organizational Behaviour, Stephen .P. Robbins, 1st Edition, Pearson, 2013
4. Brilliant- Communication skills, Gill Hasson, 1st Edition, Pearson Life, 2011
5. The Ace of Soft Skills: Attitude, Communication and Etiquette for success, Gopala Swamy Ramesh, 5th Edition, Pearson, 2013
6. Developing your influencing skills, Deborah Dalley, Lois Burton, Margaret, Green hall, 1st Edition Universe of Learning LTD, 2010
7. Communication skills for professionals, Konar nira, 2nd Edition, New arrivals – PHI, 2011
8. Personality development and soft skills, Barun K Mitra, 1st Edition, Oxford Press, 2011
9. Soft skill for everyone, Butter Field, 1st Edition, Cengage Learning india pvt.ltd, 2011
10. Soft skills and professional communication, Francis Peters SJ, 1st Edition, Mc Graw Hill Education, 2011
11. Effective communication, John Adair, 4th Edition, Pan Mac Millan, 2009
12. Bringing out the best in people, Aubrey Daniels, 2nd Edition, Mc Graw Hill, 1999

Theory Sessional examination syllabus

Sessional No.	Syllabus Chapter No.
I	1 to 5
II	6 to 10

BP106RM T - REMEDIAL MATHEMATICS (THEORY)

Teacher/s: Mr. D.H.P. Gowda (DHP)

30 Hours (2 Hrs/ week)

Scope: This is an introductory course in mathematics. This subject deals with the introduction to Partial fraction, Logarithm, matrices and Determinant, Analytical geometry, Calculus, differential equation and Laplace transform.

Objectives Upon completion of the course the student shall be able to:

1. solve problems pertaining to Trigonometry, Analytical geometry, matrices, determinants, integration, differential equation, Laplace transform
2. explain importance of mathematics in pharmacy

Course materials:

Recommended Books (Latest Edition)

1. Differential Calculus by Shanthinarayan
2. Pharmaceutical Mathematics with application to Pharmacy by Panchaksharappa Gowda D.H.
3. Integral Calculus by Shanthinarayan
4. Higher Engineering Mathematics by Dr.B.S.Grewal

Lecture wise programme:

	Topic	Hours
1	Partial fraction Introduction, Polynomial, Rational fractions, Proper and Improper fractions, Partial fraction , Resolving into Partial fraction, Application of Partial Fraction in Chemical Kinetics and Pharmacokinetics	02
2	Logarithms Introduction, Definition, Theorems/ properties of logarithms, Common logarithms, Characteristic and Mantissa, worked examples, application of logarithm to solve pharmaceutical problems.	02
3	Matrices and Determinant Introduction to matrices, Types of matrices, Operation on matrices, Transpose of a matrix, Matrix Multiplication, Determinants, Properties of determinants , Product of determinants, Minors and co-Factors, Adjoint or adjugate of a square matrix , Singular and non-singular matrices, Inverse of a matrix, Solution of system of linear of equations using matrix method, Cramer's rule, Characteristic equation and roots of a square matrix, Cayley – Hamilton theorem, Application of Matrices in solving Pharmacokinetic equations	07
4	Analytical Geometry Introduction: Signs of the Coordinates, Distance formula,	04

	Straight Line : Slope or gradient of a straight line, Conditions for parallelism and perpendicularity of two lines, Slope of a line joining two points, Slope – intercept form of a straight line	
5	Calculus Differentiation : Introductions, Derivative of a function, Derivative of a constant, Derivative of a product of a constant and a function, Derivative of the sum or difference of two functions, Derivative of the product of two functions (product formula), Derivative of the quotient of two functions (Quotient formula) – Without Proof , Derivative of x^n w.r.t. where n is any rational number, Derivative of e^x , Derivative of $\log_e x$, Derivative of a^x , Derivative of trigonometric functions from first principles (without Proof), Successive Differentiation, Conditions for a function to be a maximum or a minimum at a point. Applications Integration: Introduction, Definition, Standard formulae, Rules of integration, Method of substitution, Method of Partial fractions, Integration by parts, definite integrals, application	10
6	Differential Equations : Some basic definitions, Order and degree, Equations in separable form, Homogeneous equations, Linear Differential equations, Exact equations, Application in solving Pharmacokinetic equations	03
7	Laplace Transform : Introduction, Definition, Properties of Laplace transform, Laplace Transforms of elementary functions, Inverse Laplace transforms, Laplace transform of derivatives, Application to solve Linear differential equations, Application in solving Chemical kinetics and Pharmacokinetics equations	02

Theory Sessional examination syllabus

Sessional No.	Syllabus Chapter No.
I	1 to 4
II	5 to 7

BP106RBT - REMEDIAL BIOLOGY (THEORY)

Teacher/s: Mr Vageesh Revadigar (VR)

30 Hours (2 Hrs/ week)

Scope: To learn and understand the components of living world, structure and functional system of plant and animal kingdom.

Objectives: Upon completion of the course, the student shall be able to

- know the classification and salient features of five kingdoms of life
- understand the basic components of anatomy & physiology of plant
- know understand the basic components of anatomy & physiology animal with special reference to human

Lecture wise programme:

	Topic	Hours
1	<p>Living World:</p> <ul style="list-style-type: none"> • Definition and characters of living organisms • Diversity in the living world • Binomial nomenclature • Five kingdoms classification. Salient features of Monera, Protista, Fungi, Animalia and Plantae, Virus, <p>Morphology and Anatomy of Flowering plants</p> <ul style="list-style-type: none"> • Morphology of different parts of flowering plants – Root, stem, leaf inflorescence, flower, fruit, seed. • General Anatomy of Root, stem, leaf of monocotyledons & Dicotyledonous Plant. 	05
2	<p>Plants and mineral nutrition:</p> <ul style="list-style-type: none"> • Essential mineral, macro and micronutrients • Nitrogen metabolism, Nitrogen cycle, biological nitrogen fixation <p>Photosynthesis</p> <ul style="list-style-type: none"> • Autotrophic nutrition, photosynthesis, Photosynthetic pigments, Factors affecting photosynthesis. <p>Plant respiration:</p> <ul style="list-style-type: none"> • Cellular respiration, glycolysis, fermentation (anaerobic) 	05
3	<p>Plant growth and development</p> <ul style="list-style-type: none"> • Phases and rate of plant growth, Condition of growth, • Introduction to plant growth regulators <p>Cell- The unit of life</p> <ul style="list-style-type: none"> • Structure and functions of cell and cell organelles, cell division <p>Tissues</p> <ul style="list-style-type: none"> • Definition, types of tissues, location and functions. 	05

- 4 Body fluids and circulation** **08**
- Composition of blood, blood groups, coagulation of blood
 - Composition and functions of lymph
 - Human circulatory system
 - Structure of human heart and blood vessels
 - Cardiac cycle, cardiac output and ECG
- Digestion and Absorption**
- Human alimentary canal and digestive glands
 - Role of digestive enzymes
 - Digestion, absorption and assimilation of digested food
- Breathing and respiration**
- Human respiratory system
 - Mechanism of breathing and its regulation
 - Exchange of gases, transport of gases and regulation of respiration
 - Respiratory volumes
- 5 Excretory products and their elimination** **07**
- Modes of excretion
 - Human excretory system- structure and function
 - Urine formation
 - Renin angiotensin system
- Neural control and coordination**
- Definition and classification of nervous system
 - Structure of a neuron
 - Generation and conduction of nerve impulse
 - Structure of brain and spinal cord
 - Functions of cerebrum, cerebellum, hypothalamus and medulla oblongata
- Chemical coordination and regulation**
- Endocrine glands and their secretions
 - Functions of hormones secreted by endocrine glands
- Human reproduction**
- Parts of female reproductive system
 - Parts of male reproductive system
 - Spermatogenesis and Oogenesis
 - Menstrual cycle

Text books

- a. Text book of Biology by S. B. Gokhale
- b. A Text book of Biology by Dr. Thulajappa and Dr. Seetaram.

Reference books

- a. A Text book of Biology by B.V. Sreenivasa Naidu
- b. A Text book of Biology by Naidu and Murthy
- c. Botany for Degree students By A.C.Dutta.
- d. Outlines of Zoology by M. Ekambaranatha ayyer and T. N. Ananthakrishnan.
- e. A manual for pharmaceutical biology practical by S. B. Gokhale and C. K. Kokate

Theory Sessional examination syllabus

Sessional No.	Syllabus Chapter No.
I	1, 2, 3
II	4 & 5

BP112P – REMEDIAL BIOLOGY (PRACTICALS)**Teacher/s: Mr Vageesh Revadigar (VR)****30 Hours (2 Hrs/ week)**

1. Introduction to experiments in biology
 - a) Study of Microscope
 - b) Section cutting techniques
 - c) Mounting and staining
 - d) Permanent slide preparation
2. Study of cell and its inclusions
3. Study of Stem, Root, Leaf, seed, fruit, flower and their modifications
4. Detailed study of frog by using computer models
5. Microscopic study and identification of tissues pertinent to Stem, Root, Leaf, seed, fruit and flower
6. Identification of bones
7. Determination of blood group
8. Determination of blood pressure
9. Determination of tidal volume

Reference books

1. Practical human anatomy and physiology. by S.R.Kale and R.R.Kale.
2. A Manual of pharmaceutical biology practical. By S.B.Gokhale .C.k.Kokate. S.P.Shriwastava.
3. Biology practical manual according to National core curriculum .Biology forum of Karnataka prof .M.J.H.Shafi

JSS Academy of Higher Education & Research
JSS College of Pharmacy
 Sri Shivarathreeswara Nagara, Mysore-570015
 CLASSTIME TABLE – 2020-21

Class: B. PHARM (Semester- I)

Lunch Break: 1.00 to 2.00 PM
 Tea Break: 10.40 to 11.10 AM
 3.50 PM to 4.05 PM

Time Day	9.00-9.50AM	9.50-10.40AM	11.10-12.05PM	12.05-1.00PM	2.00-2.55PM	2.55-3.50PM	4.05-5.00PM	5.00-5.55PM
Monday	Communication Skills	Communication Skills	Pharmaceutical Inorg. Chem. DAK	Pharmaceutical Analysis SRD	Human Anatomy & Physiology AMM	Pharmaceutics ST	Pharmaceutical Inorg. Chem. DAK	-----
Tuesday	Pharmaceutics MB	Human Anatomy & Physiology AMM	Pharmaceutics ST(Tu)	Pharmaceutical Inorg. Chem. DAK	← Human Anatomy & Physiology ← Pharmaceutical Analysis ← Pharm. Inorganic Chemistry ← Pharmaceutics		-----Batch - I --AMM----- -----Batch - II ----SUB----- -----Batch - III--DAK----- -----Batch - IV---MB-----	
Wednesday	Pharmaceutics MB	Pharmaceutical Analysis SRD	Human Anatomy & Physiology AMM	Pharmaceutical Analysis (Tu) SRD	← Human Anatomy & Physiology ← Pharmaceutical Analysis ← Pharm. Inorganic Chemistry ← Pharmaceutics		-----Batch - II --AMM----- -----Batch - III----SRD----- -----Batch - IV---DAK----- -----Batch - I ----ST-----	
Thursday	← Remedial Biology--Batch -HP→ ← Communication Skills--Batch I→		Remedial Biology HP	Remedial Biology HP	← Human Anatomy & Physiology ← Pharmaceutical Analysis ← Pharm. Inorganic Chemistry ← Pharmaceutics		-----Batch - III---AMM----- -----Batch - IV---SRD----- -----Batch - I ---DAK----- -----Batch - II --ST-----	
	-----	Remedial Maths DHP	Remedial Maths DHP	← Communication Skills--Batch II→				
Friday	← Communication Skills --Batch-IV→		← Communication Skills - Batch- III→		← Human Anatomy & Physiology ← Pharmaceutical Analysis ← Pharm. Inorganic Chemistry ← Pharmaceutics		-----Batch - IV---AMM----- -----Batch - I---SUB----- -----Batch - II---DAK----- -----Batch - III -BM-----	
Saturday	Pharmaceutical Analysis SRD	Human Anatomy & Physiology(Tu) AMM	Pharmaceutical Inorg. Chem. (Tu) DAK	-----	-----			

*Effective from: 8th July 2020

Note: 1. No tea break for practical's

Time table Coordinator

Copy: SNB/LNB/SCF/e-copy-Teachers/ Office in charge/Time table / Time table Coordinator

Principal

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